

Homonuclear Diatomic Molecule

Energy Diagram

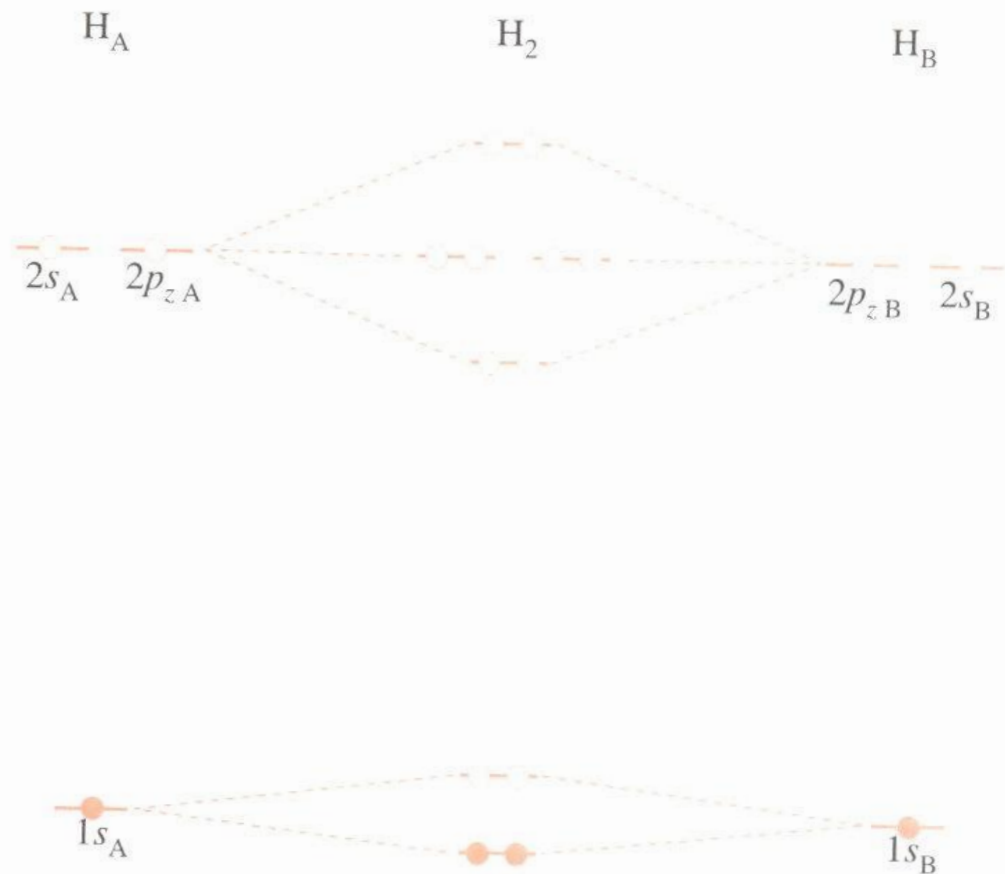
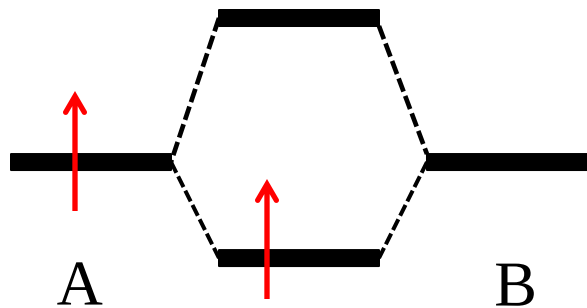


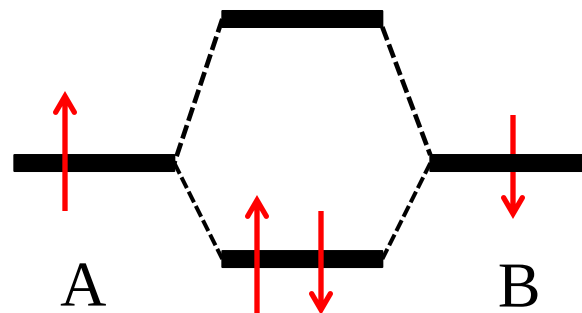
FIGURE 10.28

The six spatial molecular orbitals that are obtained when the LCAO-MO is a linear combination of six atomic orbitals, as in Equation 10.55. Only the molecular orbital of the lowest energy is occupied in the ground electronic state of H₂. The five unoccupied orbitals are called virtual orbitals.

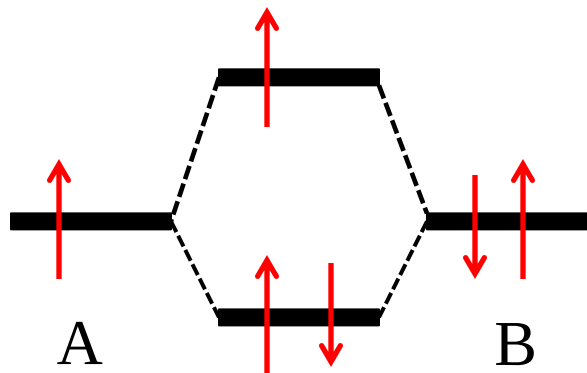
H_2^+ , H_2 , He_2^+ , He_2



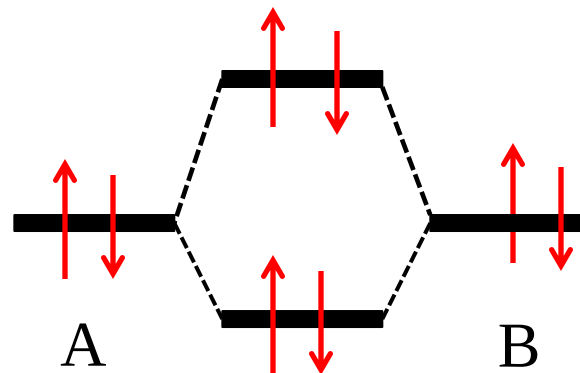
H_2^+ $R_{\text{eq}} =$ Angstrom
Binding Energy kcal/mol



H_2 $R_{\text{eq}} =$ Angstrom
Binding Energy kcal/mol

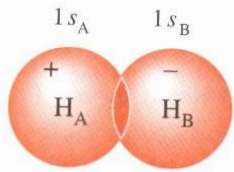


He_2^+ $R_{\text{eq}} =$ Angstrom
Binding Energy kcal/mol



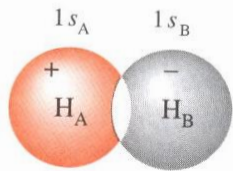
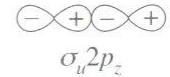
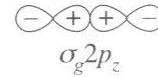
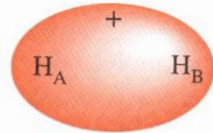
He_2 $R_{\text{eq}} =$ Angstrom
Binding Energy Kcal mol

Molecular Orbitals: Sum of Atomic Orbitals Sigma and Pi



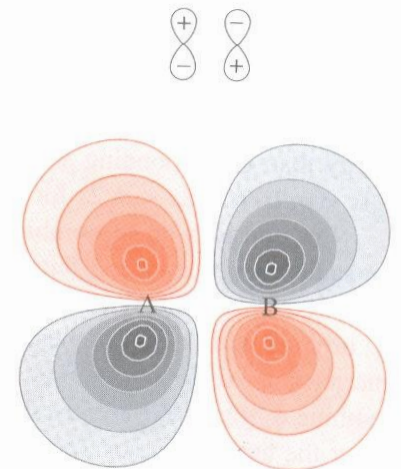
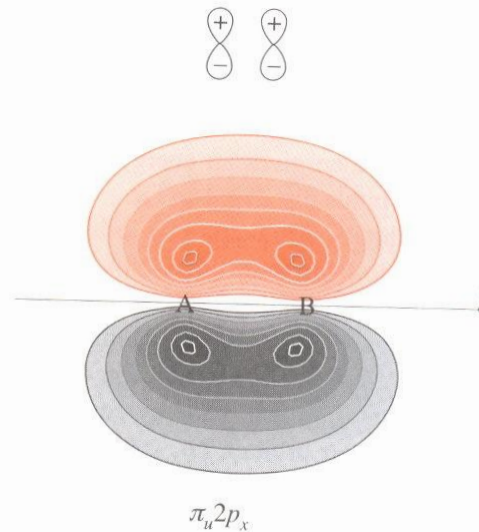
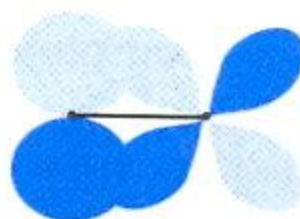
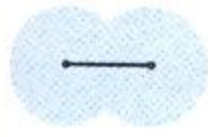
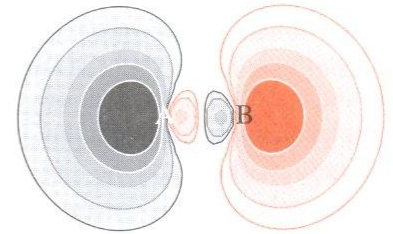
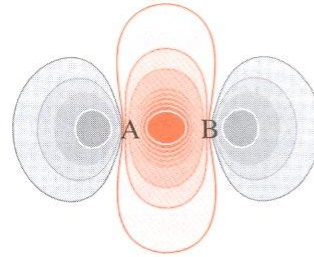
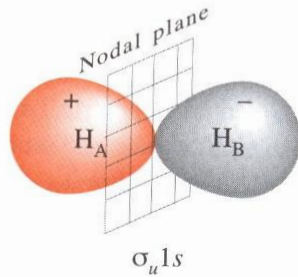
$$1s_A + 1s_B$$

=



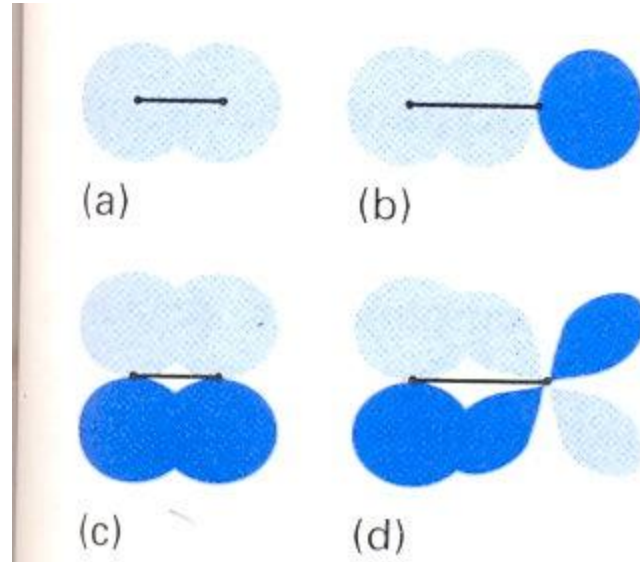
$$1s_A - 1s_B$$

=

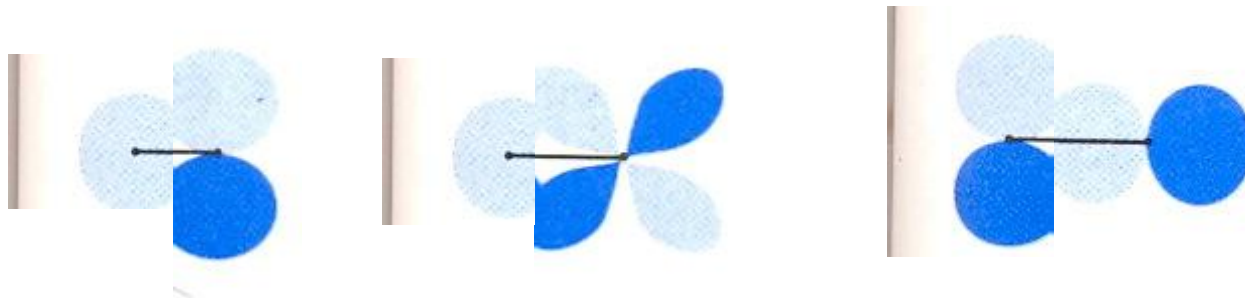


Symmetry of Orbitals Important

Allowed Combinations



Not Allowed Combinations: $S=0$, $H_{AB}=0$



Binding Energy: Orbital Interaction

- Binding depends on overlap of atomic orbitals:

Energies of atomic orbital closer the better

Consider the interaction of Orbital of atom A and B at E_A and E_B

$$\begin{vmatrix} H_{AA} - E & H_{AB} - ES \\ H_{BA} - ES & H_{BB} - E \end{vmatrix} = \begin{vmatrix} \alpha_A - E & \beta - ES \\ \beta - ES & \alpha_B - E \end{vmatrix} = 0$$

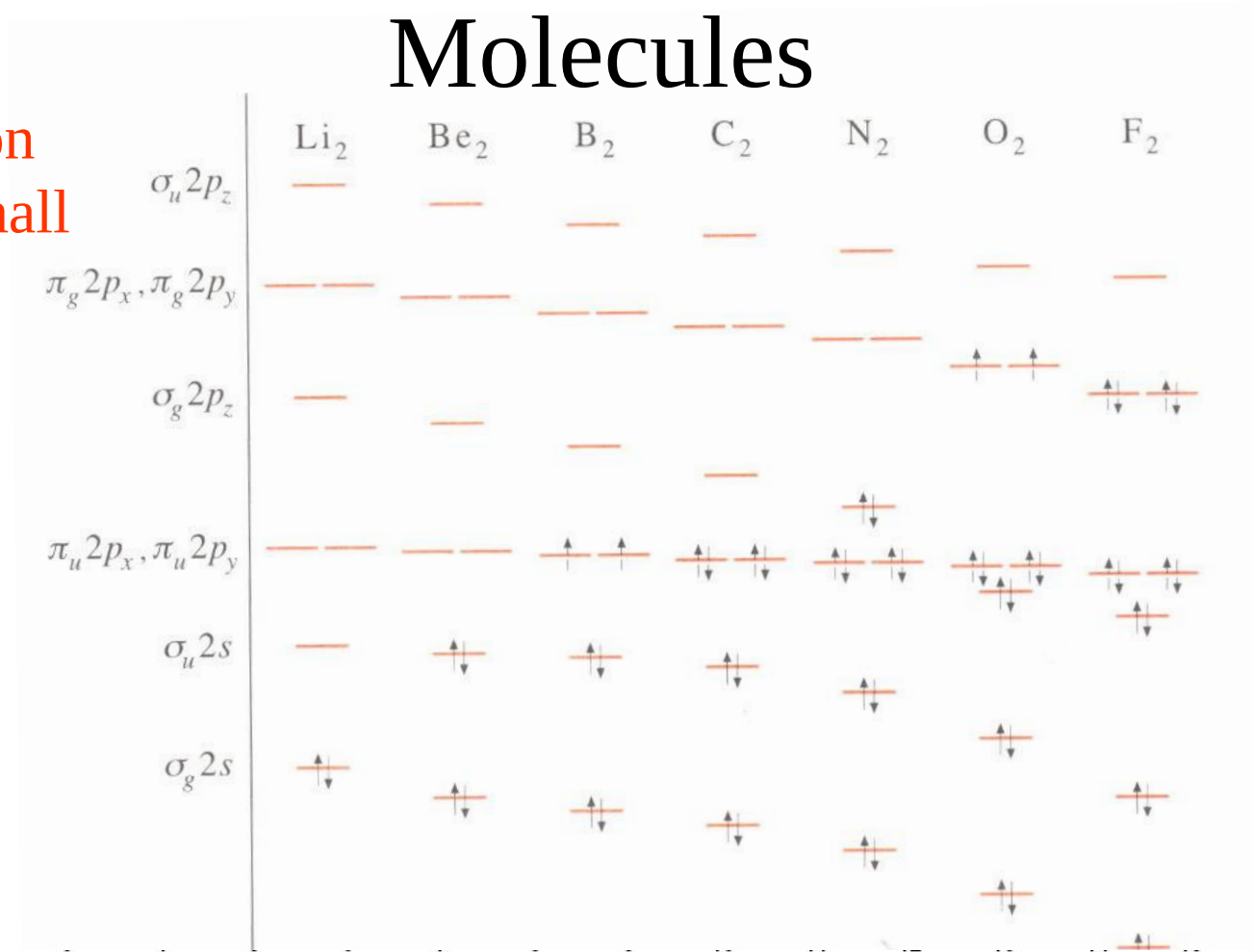
Assume $S=0$ and $|E_A - E_B| \gg E_{AB}$

$$E_+ = \alpha_A - \frac{\beta^2}{\alpha_B - \alpha_A}; \quad E_- = \alpha_B + \frac{\beta^2}{\alpha_B - \alpha_A}$$

Second Row Homonuclear diatomic Molecules

Ignore
1S interaction
Since too small

Valance
Orbitals are
considered

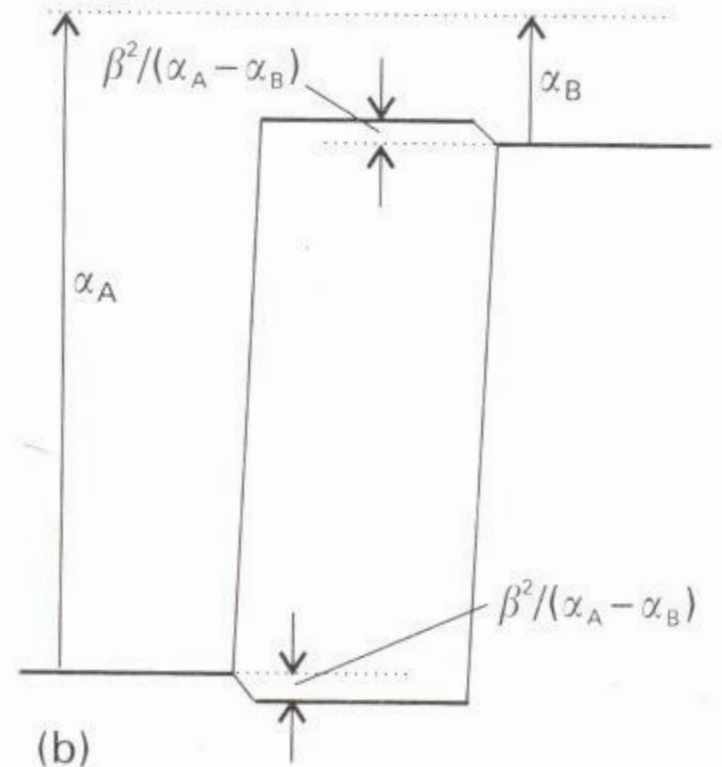
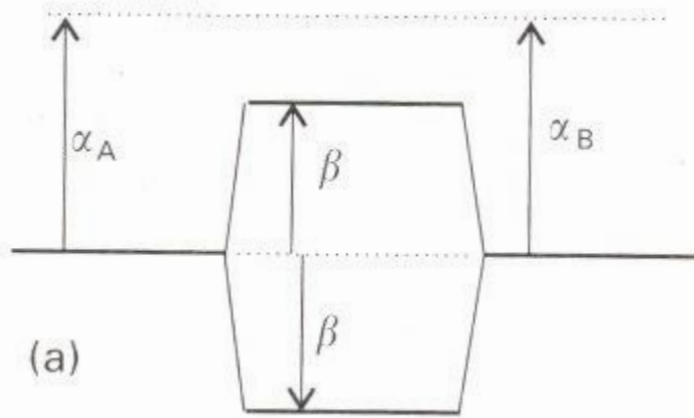


HOMO LUMO Gap $\leftrightarrow$ Band Gap

hydrogen 1 H 1.0079	beryllium 4 Be 9.0122
lithium 3 Li 6.941	magnesium 12 Mg 24.305

boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180
aluminum 13 Al 26.981	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.06	chlorine 17 Cl 35.45	argon 18 Ar 39.948

Homo/Heteronuclear Diatomic Molecules



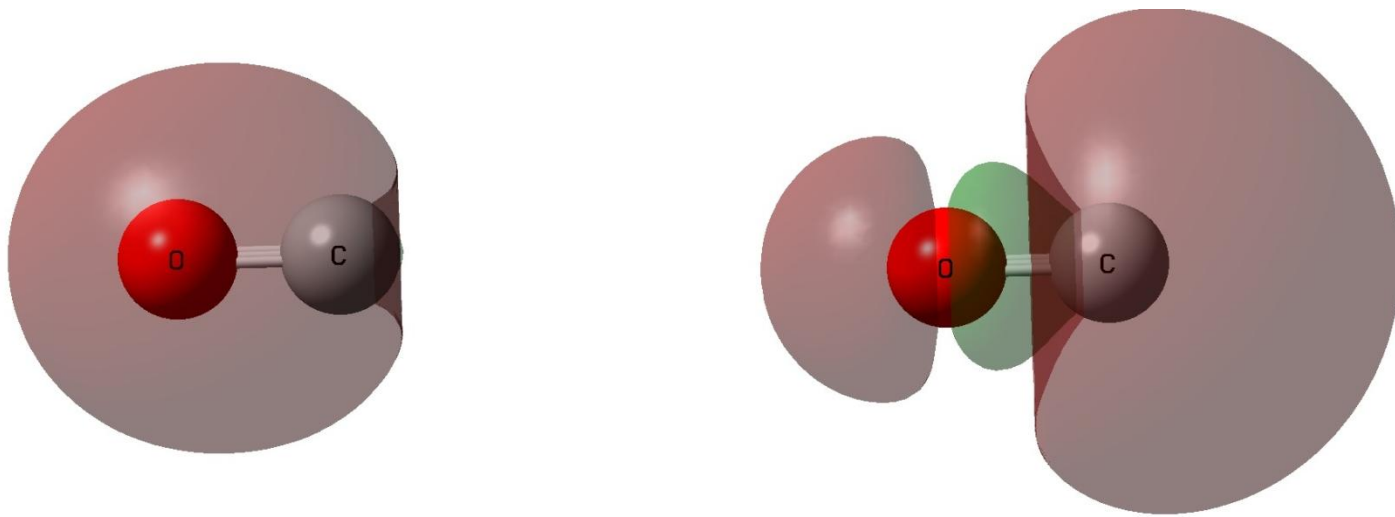
CO Molecule: Orbitals

In heterodiatomic molecules
electronegativity of the atoms determine the
shape of the orbital

$$\Psi = C_A \psi_A + C_B \psi_B$$

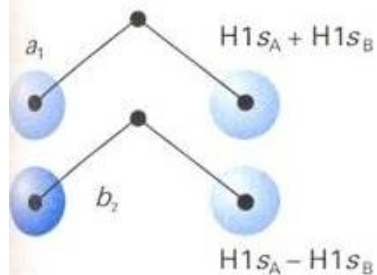
C_A and C_B define the density on A and B

If A is more electronegative:

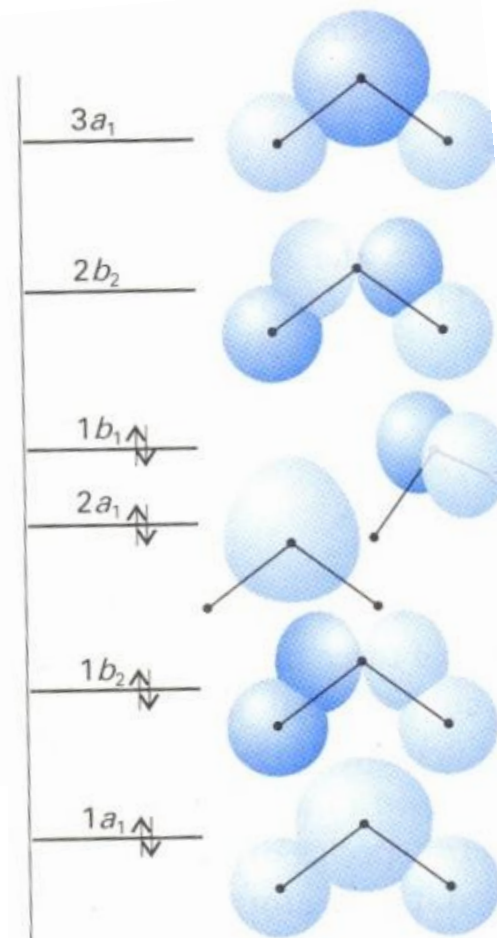


Polyatomic Molecules

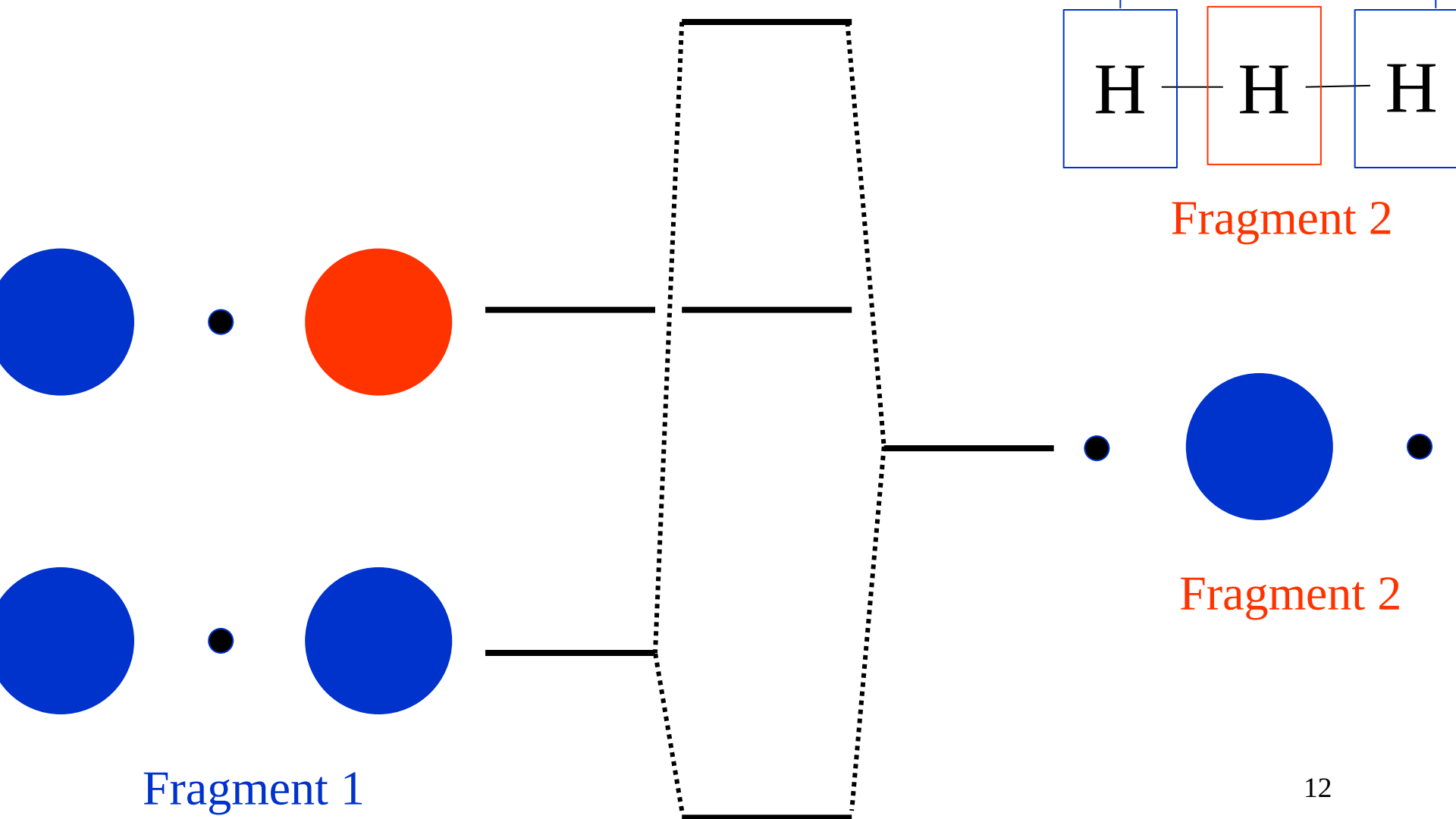
Water H₂O



C_{2v}	O2s	O2p _x	O2p _y	O2p _z	H1s _A	H1s _B
E	O2s	O2p _x	O2p _y	O2p _z	H1s _A	H1s _B
C_2	O2s	-O2p _x	-O2p _y	O2p _z	H1s _B	H1s _A
σ_v	O2s	O2p _x	-O2p _y	O2p _z	H1s _B	H1s _A
σ'_v	O2s	-O2p _x	O2p _y	O2p _z	H1s _A	H1s _B



Linear H_3



Hartree-Fock Roothaan Equation Electronic Structure

Kaito Takahashi

Born-Oppenheimer Approximation

Solve for the electron at a fixed nuclear geometry

$$\hat{H}^0(\mathbf{r}; \mathbf{R}) \Psi_n^{el}(\mathbf{r}; \mathbf{R}) = \left[-\frac{1}{2} \sum_{i=1}^n \nabla_i^2 + V(\mathbf{r}, \mathbf{R}) \right] \Psi_n^{el}(\mathbf{r}; \mathbf{R}) = E_n(\mathbf{R}) \Psi_n^{el}(\mathbf{r}; \mathbf{R})$$

Calculate many nuclear geometries to obtain the potential energy surface

$$\hat{H}_{NU}(\mathbf{R}) \chi_{n, \mathbf{v}_n}^{NU}(\mathbf{R}) = E_{el, NU} \chi_{n, \mathbf{v}_n}^{NU}(\mathbf{R})$$

$$\hat{H}_{NU}(\mathbf{R}) = \left(-\frac{1}{2} \sum_{I=1}^N \frac{1}{M_I} \nabla_I^2 + V(\mathbf{R}) \right)$$

Problem To Solve with trial WF

$$\hat{H}^0(\mathbf{r}; \mathbf{R}) \Psi_n^{el}(\mathbf{r}; \mathbf{R})$$

$$= \left[-\frac{1}{2} \sum_{i=1}^n \nabla_i^2 + \sum_{I=1}^N \sum_{J \neq I}^N \frac{Z_I Z_J}{|\mathbf{R}_I - \mathbf{R}_J|} - \sum_{I=1}^N \sum_{i=1}^n \frac{Z_I}{|\mathbf{R}_I - \mathbf{r}_i|} + \sum_{i=1}^n \sum_{j \neq i}^n \frac{1}{|\mathbf{r}_i - \mathbf{r}_j|} \right] \Psi_n^{el}(\mathbf{r}; \mathbf{R})$$

$$= E_n(\mathbf{R}) \Psi_n^{el}(\mathbf{r}; \mathbf{R})$$

$$\Psi(\mathbf{x}_1, \mathbf{x}_2, \dots, \mathbf{x}_n) = \frac{1}{\sqrt{n!}} \begin{vmatrix} \psi_i(\mathbf{x}_1) & \psi_j(\mathbf{x}_1) & \dots & \psi_k(\mathbf{x}_1) \\ \psi_i(\mathbf{x}_2) & \psi_j(\mathbf{x}_2) & \dots & \psi_k(\mathbf{x}_2) \\ \dots & \dots & \dots & \dots \\ \psi_i(\mathbf{x}_n) & \psi_j(\mathbf{x}_n) & \dots & \psi_k(\mathbf{x}_n) \end{vmatrix}$$

Operators and Matrix Elements

Hamiltonian is a sum of one and two electron operators
For example hydrogen molecule

$$\left[-\frac{1}{2} \nabla_1^2 - \frac{1}{|\mathbf{r}_{1A}|} - \frac{1}{|\mathbf{r}_{1B}|} - \frac{1}{2} \nabla_2^2 - \frac{1}{|\mathbf{r}_{2A}|} - \frac{1}{|\mathbf{r}_{2B}|} + \left[\frac{1}{|\mathbf{R}|} + \frac{1}{|\mathbf{r}_{12}|} \right] \right]$$

$$= \left[h_1 + h_2 + \frac{1}{|\mathbf{R}|} + \frac{1}{|\mathbf{r}_{12}|} \right]$$

$$h_1, h_2 : \quad \frac{1}{|\mathbf{r}_{12}|} :$$

$$\Psi_0(\mathbf{x}_1, \mathbf{x}_2) = \frac{1}{\sqrt{2}} (\psi_i(\mathbf{x}_1) \psi_j(\mathbf{x}_2) - \psi_j(\mathbf{x}_1) \psi_i(\mathbf{x}_2))$$

One Electron Operator Matrix Element

$$\begin{aligned}
 \langle 0|h_1|0\rangle &= 2^{-1} \iint d\mathbf{x}_1 d\mathbf{x}_2 \left(\psi_i(\mathbf{x}_1) \psi_j(\mathbf{x}_2) - \psi_j(\mathbf{x}_1) \psi_i(\mathbf{x}_2) \right)^* \\
 &\quad \times h(\mathbf{r}_1) \left(\psi_i(\mathbf{x}_1) \psi_j(\mathbf{x}_2) - \psi_j(\mathbf{x}_1) \psi_i(\mathbf{x}_2) \right) \\
 &= 2^{-1} \iint d\mathbf{x}_1 d\mathbf{x}_2 [\\
 &\hspace{15em}]
 \end{aligned}$$

Due to orthonormality of the spin orbitals last two are zero with integration with respect to $\mathbf{x}_2$

One Electron Operator 2

$$\begin{aligned}\langle 0|h_1|0\rangle &= 2^{-1} \iint d\mathbf{x}_1 d\mathbf{x}_2 \left[\psi_i^*(\mathbf{x}_1) \psi_j^*(\mathbf{x}_2) h(\mathbf{r}_1) \psi_i(\mathbf{x}_1) \psi_j(\mathbf{x}_2) \right. \\ &\quad \left. + \psi_j^*(\mathbf{x}_1) \psi_i^*(\mathbf{x}_2) h(\mathbf{r}_1) \psi_j(\mathbf{x}_1) \psi_i(\mathbf{x}_2) \right] \\ &= 2^{-1} \int \psi_i^*(\mathbf{x}_1) h(\mathbf{r}_1) \psi_i(\mathbf{x}_1) d\mathbf{x}_1 + 2^{-1} \int \psi_j^*(\mathbf{x}_1) h(\mathbf{r}_1) \psi_j(\mathbf{x}_1) d\mathbf{x}_1 \\ &= 2^{-1} \left(\langle \psi_i | h | \psi_i \rangle + \langle \psi_j | h | \psi_j \rangle \right) \\ \langle 0|h_2|0\rangle &= 2^{-1} \left(\langle \psi_i | h | \psi_i \rangle + \langle \psi_j | h | \psi_j \rangle \right)\end{aligned}$$

Two Electron Operator Matrix Element

$$\begin{aligned}
 \left\langle 0 \left| |\mathbf{r}_{12}|^{-1} \right| 0 \right\rangle &= 2^{-1} \iint d\mathbf{x}_1 d\mathbf{x}_2 \left(\psi_i(\mathbf{x}_1) \psi_j(\mathbf{x}_2) - \psi_j(\mathbf{x}_1) \psi_i(\mathbf{x}_2) \right)^* \\
 &\quad \times |\mathbf{r}_{12}|^{-1} \left(\psi_i(\mathbf{x}_1) \psi_j(\mathbf{x}_2) - \psi_j(\mathbf{x}_1) \psi_i(\mathbf{x}_2) \right) \\
 &= 2^{-1} \iint d\mathbf{x}_1 d\mathbf{x}_2 [
 \end{aligned}$$

Two Electron Operator 2

$$\begin{aligned}\langle 0 | |\mathbf{r}_{12}|^{-1} | 0 \rangle &= \iint d\mathbf{x}_1 d\mathbf{x}_2 \left[\psi_i^*(\mathbf{x}_1) \psi_j^*(\mathbf{x}_2) |\mathbf{r}_{12}|^{-1} \psi_i(\mathbf{x}_1) \psi_j(\mathbf{x}_2) \right. \\ &\quad \left. - \psi_i^*(\mathbf{x}_1) \psi_j^*(\mathbf{x}_2) |\mathbf{r}_{12}|^{-1} \psi_j(\mathbf{x}_1) \psi_i(\mathbf{x}_2) \right] \\ &= \langle ij | ij \rangle - \langle ij | ji \rangle\end{aligned}$$

$$\langle ij | kl \rangle = \langle ji | lk \rangle = \langle kl | ij \rangle^* = \langle lk | ji \rangle^*$$

Symbols physics people use: $\langle ij || kl \rangle = \langle ij | kl \rangle - \langle ij | lk \rangle$

Symbols chemistry people use:

$$[ij | kl] = \langle ik | jl \rangle = \iint d\mathbf{x}_1 d\mathbf{x}_2 \left[\psi_i^*(\mathbf{x}_1) \psi_j(\mathbf{x}_1) |\mathbf{r}_{12}|^{-1} \psi_k^*(\mathbf{x}_2) \psi_l(\mathbf{x}_2) \right]$$

So the expectation value of the hamiltonian with the Slater determinant for two electron system is

Expectation Value of Slater Det

$$|\psi_1\psi_2\cdots\psi_n\rangle = \frac{1}{\sqrt{n!}} \begin{vmatrix} \psi_i(\mathbf{x}_1) & \psi_j(\mathbf{x}_1) & \cdots & \psi_n(\mathbf{x}_1) \\ \psi_i(\mathbf{x}_2) & \psi_j(\mathbf{x}_2) & \cdots & \psi_n(\mathbf{x}_2) \\ \cdots & \cdots & \cdots & \cdots \\ \psi_i(\mathbf{x}_n) & \psi_j(\mathbf{x}_n) & \cdots & \psi_n(\mathbf{x}_n) \end{vmatrix}$$

For one electron operators we consider the sum of each electron, for two electron operators the sum of pairs of electrons

two electron system Hamiltonian was $\left[h_1 + h_2 + \frac{1}{|\mathbf{R}|} + \frac{1}{|\mathbf{r}_{12}|} \right]$

one electron operators for n electron system $\sum_{i=1}^n h_i$

two electron operator for n electron system $\sum_{i=1}^n \sum_{j \neq i}^n |\mathbf{r}_{ij}|^{-1}$

Expectation of One Electron Operator of Slater Det

$$\left\langle \psi_1 \psi_2 \dots \psi_n \left| \sum_{i=1}^n h_i \right| \psi_1 \psi_2 \dots \psi_n \right\rangle = \left\langle \psi_1 \psi_2 \dots \psi_n \left| h_1 + h_2 + \dots h_n \right| \psi_1 \psi_2 \dots \psi_n \right\rangle$$

Since electron in slater determinant is indistinguishable we consider for just electron 1 and multiply by the number of electrons

$$\begin{aligned} \left\langle \psi_1 \psi_2 \dots \psi_n \left| \sum_{i=1}^n h_i \right| \psi_1 \psi_2 \dots \psi_n \right\rangle &= n \left\langle \psi_1 \psi_2 \dots \psi_n \left| h_1 \right| \psi_1 \psi_2 \dots \psi_n \right\rangle \\ &= \sum_{i=1}^n \langle \psi_i | h_1 | \psi_i \rangle \end{aligned}$$

Expectation Value of a Two electron Operator for Slater Det

$$\left\langle \psi_1 \psi_2 \dots \psi_n \left| \sum_{i=1}^n \sum_{j \neq i}^n |\mathbf{r}_{ij}|^{-1} \right| \psi_1 \psi_2 \dots \psi_n \right\rangle$$

$$= \left\langle \psi_1 \psi_2 \dots \psi_n \left| |\mathbf{r}_{12}|^{-1} + |\mathbf{r}_{13}|^{-1} + \dots + |\mathbf{r}_{n-1n}|^{-1} \right| \psi_1 \psi_2 \dots \psi_n \right\rangle$$

Since electron in slater determinant is indistinguishable we solve for electron 1 and 2 and multiply by the number of pairs

Hartree Fock Approximation 1

$$\hat{H}^0(\mathbf{r}; \mathbf{R}) = \sum_{i=1}^n h_i + \sum_{i=1}^n \sum_{j \neq i}^n \frac{1}{|\mathbf{r}_i - \mathbf{r}_j|}$$

$$|\psi_1 \psi_2 \dots \psi_n\rangle = \frac{1}{\sqrt{n!}} \begin{vmatrix} \psi_i(\mathbf{x}_1) & \psi_j(\mathbf{x}_1) & \dots & \psi_n(\mathbf{x}_1) \\ \psi_i(\mathbf{x}_2) & \psi_j(\mathbf{x}_2) & \dots & \psi_n(\mathbf{x}_2) \\ \dots & \dots & \dots & \dots \\ \psi_i(\mathbf{x}_n) & \psi_j(\mathbf{x}_n) & \dots & \psi_n(\mathbf{x}_n) \end{vmatrix}$$

$$\langle \psi_1 \psi_2 \dots \psi_n | \hat{H}^0 | \psi_1 \psi_2 \dots \psi_n \rangle = \sum_{i=1}^n \langle \psi_i | h_i | \psi_i \rangle + \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle ij | ij \rangle - \langle ij | ji \rangle)$$

$$\langle \psi_i | \psi_j \rangle = \delta_{ij}$$

How do you find the best answer for the spin orbitals? 24

Hartree Fock Approximation 2

Functional minimization with constraint

Functional derivative with respect to change of spin orbital

$\psi_i \rightarrow \psi_i + \delta\psi_i$ should be zero

$$\mathfrak{I}(\{\psi_i + \delta\psi_i\}) - \mathfrak{I}(\{\psi_i\}) = \delta\mathfrak{I}(\{\psi_i\}) = 0$$

$$\delta\langle\psi_i|\psi_j\rangle = \langle\delta\psi_i|\psi_j\rangle + \langle\psi_i|\delta\psi_j\rangle$$

Hartree Fock Approximation 3

$$\begin{aligned} \delta \langle \psi_1 \psi_2 \dots \psi_n | \hat{H}^0 | \psi_1 \psi_2 \dots \psi_n \rangle &= \sum_{i=1}^n \langle \delta \psi_i | h_1 | \psi_i \rangle + \sum_{i=1}^n \langle \psi_i | h_1 | \delta \psi_i \rangle \\ &+ \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle \delta i j | i j \rangle - \langle \delta i j | j i \rangle) + \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle i \delta j | i j \rangle - \langle i \delta j | j i \rangle) \\ &+ \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle i j | \delta i j \rangle - \langle i j | \delta j i \rangle) + \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle i j | i \delta j \rangle - \langle i j | j \delta i \rangle) \end{aligned}$$

Remember the exchange relationship of two electron integral

$$\langle i j | k l \rangle = \langle j i | l k \rangle = \langle k l | i j \rangle^* = \langle l k | j i \rangle^*$$

$$\frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle i \delta j | i j \rangle - \langle i \delta j | j i \rangle) =$$

=

Hartree Fock Approximation 4

$$\frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle ij | \delta_{ij} \rangle - \langle ij | \delta_{ji} \rangle) =$$

=

$$\frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle ij | i \delta_j \rangle - \langle ij | j \delta_i \rangle) = \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle i \delta_j | ij \rangle^* - \langle j \delta_i | ij \rangle^*)$$

$$= \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle \delta_j i | ji \rangle^* - \langle \delta_i j | ji \rangle^*)$$

$$= \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle \delta_i j | ij \rangle^* - \langle \delta_i j | ji \rangle^*)$$

Hartree Fock Approximation 5

$$\delta\mathfrak{S}(\{\psi_i\}) = \delta\left\langle\psi_1\psi_2...\psi_n\left|\hat{H}^0\right|\psi_1\psi_2...\psi_n\right\rangle - \sum_{i=1}^n\sum_{j=1}^n\varepsilon_{ji}\left(\delta\left\langle\psi_i\left|\psi_j\right\rangle\right)\right) = 0$$

$$\sum_{i=1}^n\left\langle\delta\psi_i\left|h_1\right|\psi_i\right\rangle + \sum_{i=1}^n\sum_{j=1}^n\left(\left\langle\delta ij\left|ij\right\rangle - \left\langle\delta ij\left|ji\right\rangle - \varepsilon_{ji}\left\langle\delta\psi_i\left|\psi_j\right\rangle\right)\right) + cc = 0$$

$$\begin{aligned} & \sum_{i=1}^n\int\delta\psi_i^*(\mathbf{x}_1)h(\mathbf{r}_1)\psi_i(\mathbf{x}_1)d\mathbf{x}_1 + \sum_{i=1}^n\sum_{j=1}^n\iint d\mathbf{x}_1d\mathbf{x}_2\left(\delta\psi_i^*(\mathbf{x}_1)\psi_j^*(\mathbf{x}_2)\left|\mathbf{r}_{12}\right|^{-1}\psi_i(\mathbf{x}_1)\psi_j(\mathbf{x}_2)\right) \\ & - \sum_{i=1}^n\sum_{j=1}^n\iint d\mathbf{x}_1d\mathbf{x}_2\left(\delta\psi_i^*(\mathbf{x}_1)\psi_j^*(\mathbf{x}_2)\left|\mathbf{r}_{12}\right|^{-1}\psi_j(\mathbf{x}_1)\psi_i(\mathbf{x}_2)\right) - \sum_{i=1}^n\sum_{j=1}^n\varepsilon_{ji}\iint\delta\psi_i^*(\mathbf{x}_1)\psi_j(\mathbf{x}_1)d\mathbf{x}_1 \end{aligned}$$

Coulomb and Exchange Integral

Coulomb Integral

Exchange Integral

$$\begin{aligned}\langle \psi_1 \psi_2 \dots \psi_n | \hat{H}^0 | \psi_1 \psi_2 \dots \psi_n \rangle &= \sum_{i=1}^n \langle \psi_i | h_1 | \psi_i \rangle + \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle ij | ij \rangle - \langle ij | ji \rangle) \\ &= \sum_{i=1}^n h_{ii} + \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (J_{ij} - K_{ij})\end{aligned}$$

Hartree Fock Approximation 6

$$\delta\mathfrak{I}(\{\psi_i\}) = \sum_{i=1}^n \int \delta\psi_i^*(\mathbf{x}_1) \left[h(\mathbf{r}_1)\psi_i(\mathbf{x}_1) + \sum_{j=1}^n (J_j(\mathbf{x}_1) - K_j(\mathbf{x}_1))\psi_i(\mathbf{x}_1) - \sum_{j=1}^n \varepsilon_{ji}\psi_j(\mathbf{x}_1) \right] d\mathbf{x}_1 + cc$$

$$= 0$$

Since $\delta\psi_i^*(\mathbf{x}_1)$ is arbitrary [...] must be zero

Fock Operator of spin orbitals

Canonical Hartree Fock

Unitary Transform of the Hartree Fock occupied orbitals

$$\psi_i'(\mathbf{x}_1) = \sum_{j=1}^n U_{ji} \psi_j(\mathbf{x}_1) \quad i = 1, 2, \dots, n$$

$$\mathbf{U}^+ = \mathbf{U}^{-1}$$

$$\mathbf{A} = \begin{pmatrix} \psi_i(\mathbf{x}_1) & \psi_j(\mathbf{x}_1) & \dots & \psi_n(\mathbf{x}_1) \\ \psi_i(\mathbf{x}_2) & \psi_j(\mathbf{x}_2) & \dots & \psi_n(\mathbf{x}_2) \\ \dots & \dots & \dots & \dots \\ \psi_i(\mathbf{x}_n) & \psi_j(\mathbf{x}_n) & \dots & \psi_n(\mathbf{x}_n) \end{pmatrix}$$

$$\mathbf{A}' = \mathbf{A}\mathbf{U} = \begin{pmatrix} \psi_i(\mathbf{x}_1) & \psi_j(\mathbf{x}_1) & \dots & \psi_n(\mathbf{x}_1) \\ \psi_i(\mathbf{x}_2) & \psi_j(\mathbf{x}_2) & \dots & \psi_n(\mathbf{x}_2) \\ \dots & \dots & \dots & \dots \\ \psi_i(\mathbf{x}_n) & \psi_j(\mathbf{x}_n) & \dots & \psi_n(\mathbf{x}_n) \end{pmatrix} \begin{pmatrix} U_{11} & U_{12} & \dots & U_{1n} \\ U_{21} & U_{22} & \dots & U_{2n} \\ \dots & \dots & \dots & \dots \\ U_{n1} & U_{n2} & \dots & U_{nn} \end{pmatrix} = \begin{pmatrix} \psi_i'(\mathbf{x}_1) & \psi_j'(\mathbf{x}_1) & \dots & \psi_n'(\mathbf{x}_1) \\ \psi_i'(\mathbf{x}_2) & \psi_j'(\mathbf{x}_2) & \dots & \psi_n'(\mathbf{x}_2) \\ \dots & \dots & \dots & \dots \\ \psi_i'(\mathbf{x}_n) & \psi_j'(\mathbf{x}_n) & \dots & \psi_n'(\mathbf{x}_n) \end{pmatrix}$$

Canonical Hartree Fock

$$|\psi_1 \psi_2 \dots \psi_n\rangle = \frac{1}{\sqrt{n!}} \begin{vmatrix} \psi_i(\mathbf{x}_1) & \psi_j(\mathbf{x}_1) & \dots & \psi_n(\mathbf{x}_1) \\ \psi_i(\mathbf{x}_2) & \psi_j(\mathbf{x}_2) & \dots & \psi_n(\mathbf{x}_2) \\ \dots & \dots & \dots & \dots \\ \psi_i(\mathbf{x}_n) & \psi_j(\mathbf{x}_n) & \dots & \psi_n(\mathbf{x}_n) \end{vmatrix}$$

$$= \frac{1}{\sqrt{n!}} \det(\mathbf{A})$$

$$|\psi'_1 \psi'_2 \dots \psi'_n\rangle = \frac{1}{\sqrt{n!}} \det(\mathbf{A}') = \frac{1}{\sqrt{n!}} \det(\mathbf{A}\mathbf{U}) = \frac{1}{\sqrt{n!}} \det(\mathbf{A}) \det(\mathbf{U})$$

Canonical Hartree Fock

$$f(\mathbf{x}_1)\psi_i(\mathbf{x}_1) = \sum_{j=1}^n \varepsilon_{ji} \psi_j(\mathbf{x}_1) \quad i = 1, 2, \dots, n$$

Fock Operators are invariant to unitary transformation of spin orbitals

$$f(\mathbf{x}_1) = f'(\mathbf{x}_1)$$

therefore

If we find a unitary matrix that diagonalizes ε then

$$f(\mathbf{x}_1)\psi'_i(\mathbf{x}_1) = \varepsilon'_i \psi'_i(\mathbf{x}_1) \quad i = 1, 2, \dots, n$$

Canonical spin orbitals are eigen function of canonical fock operator

Orbital Energy and Total Hartree Fock Energy

$$\left[h(\mathbf{r}_1) + \sum_{j=1}^n (J_j(\mathbf{x}_1) - K_j(\mathbf{x}_1)) \right] \psi_i(\mathbf{x}_1) = \varepsilon_i \psi_i(\mathbf{x}_1) \quad i = 1, 2, \dots, n$$

$$f(1) \psi_i(\mathbf{x}_1) = \varepsilon_i \psi_i(\mathbf{x}_1) \quad i = 1, 2, \dots, n$$

ε_i : orbital energy $\varepsilon_i =$

Sum of orbital energy

$$\sum_{i=1}^n \varepsilon_i = \sum_{i=1}^n h_{ii} + \sum_{i=1}^n \sum_{j=1}^n (J_{ij} - K_{ij})$$

$$\langle \psi_1 \psi_2 \dots \psi_n | \hat{H}^0 | \psi_1 \psi_2 \dots \psi_n \rangle = \sum_{i=1}^n \langle \psi_i | h_1 | \psi_i \rangle + \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (\langle ij | ij \rangle - \langle ij | ji \rangle)$$

$$= \sum_{i=1}^n h_{ii} + \frac{1}{2} \sum_{i=1}^n \sum_{j=1}^n (J_{ij} - K_{ij})$$

Closed Shell Hartree Fock Approximation

$$\left[h(\mathbf{r}_1) + \sum_{j=1}^n (J_j(\mathbf{x}_1) - K_j(\mathbf{x}_1)) \right] \psi_i(\mathbf{x}_1) = \varepsilon_i \psi_i(\mathbf{x}_1) \quad i = 1, 2, \dots, n$$

$$f(\mathbf{r}_1) = h(\mathbf{r}_1) + \sum_{j=1}^n (J_j(\mathbf{x}_1) - K_j(\mathbf{x}_1))$$

The above equation is based on spin orbitals assuming that we will only consider closed shell systems with same orbital for different spin

$$|\psi_1 \psi_2 \dots \psi_n\rangle = |\phi_1 \bar{\phi}_1 \dots \phi_{n/2} \bar{\phi}_{n/2}\rangle$$

$$\psi_1(\mathbf{x}_1) = \phi_1(\mathbf{r}_1) \alpha(s_1) \rightarrow \phi_1$$

$$\psi_2(\mathbf{x}_1) = \phi_1(\mathbf{r}_1) \beta(s_1) \rightarrow \bar{\phi}_1$$

Closed Shell HF Approximation

$$\left[h(\mathbf{r}_1) + \sum_{b=1}^{n/2} (2J_b(\mathbf{r}_1) - K_b(\mathbf{r}_1)) \right] \phi_a(\mathbf{r}_1) = \varepsilon_a \phi_a(\mathbf{r}_1) \quad a = 1, 2, \dots, n/2$$

$$f(\mathbf{r}_1) = h(\mathbf{r}_1) + \sum_{b=1}^{n/2} (2J_b(\mathbf{r}_1) - K_b(\mathbf{r}_1))$$

The Hartree Fock approximation energy is

$$\begin{aligned} & \left\langle \phi_1 \bar{\phi}_1 \dots \phi_{n/2} \bar{\phi}_{n/2} \left| \hat{H}^0 \right| \phi_1 \bar{\phi}_1 \dots \phi_{n/2} \bar{\phi}_{n/2} \right\rangle \\ &= 2 \sum_{a=1}^{n/2} \langle \phi_a | h_1 | \phi_a \rangle + \sum_{a=1}^{n/2} \sum_{b=1}^{n/2} (2 \langle ab | ab \rangle - \langle ab | ba \rangle) \\ &= 2 \sum_{a=1}^{n/2} h_{aa} + \sum_{a=1}^{n/2} \sum_{b=1}^{n/2} (2J_{ab} - K_{ab}) \end{aligned}$$

Can use numerical basis to solve the HF equation.